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Continuing an investigation of the total alkaloids of <u>Korolkowia sewerzowii</u> Regl. [1,2], from combined fractions 25-34 we have isolated a previously unknown alkaloid (I) with the composition $C_{27}H_{43}NO_4$, mp 257-259°C (acetone), $[\alpha]_{\overline{D}} - 8$ ° (c, 0.74; ethanol).

The IR spectrum of (I) had absorption bands of active hydrogen atoms (3400 cm⁻¹) and of C=C bonds (1655 cm⁻¹). In the PMR spectrum of the base (in CD₃OD, C-60 HL), protons resonated in the form of a singlet at 0.98 ppm (3 H, 19-CH₃) and of doublets at 0.87 (3 H, 21-CH₃) and 1.04 ppm (3 H, 27-CH₃). The mass spectrum of (I) contained the main peaks of ions with m/z 445 (M)⁺, 430 (M -15)⁺, 429 (M -16)⁺ (100%), 428 (M -18)⁺ 426, 425, 414, 412, 410, 400, 358, 288, 286, 256, 180, 178, 149, 141, 129, 128 (100%), 127, 121, 114, 110.

The comparatively ready solubility of the bases in water and the presence in the mass spectrum of strong peaks of ions with m/z $(M-16)^+$ and $(M-17)^+$ permitted (I) to be assigned to the N-oxides.

When (I) was reduced with zinc and sulfuric acid, a compound identical according to its R_f value and spectral characteristics with korsine [1, 3] was obtained.

The oxidation of korsine with hydrogen peroxide led to its N-oxide, which was identical with alkaloid (I). Thus, it may be concluded unambiguously that (I) was korsine N-oxide.

LITERATURE CITED

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